

Synthesis and Antihistaminic Activity of 1-[(4-Substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines

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Received June 18, 1979, from the ^{*}Department of Chemistry, College of Pharmacy, and the [‡]Department of Experimental Medicine and Pharmacology, Faculty of Medicine, Tehran University, Tehran, Iran. Accepted for publication December 20, 1979.

Abstract □ Starting with readily available aryl 4-substituted-1,2,3-thiadiazol-5-yl ketones, a series of 1-[(4-substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines was synthesized and tested for antihistaminic and anticholinergic activities. Four compounds were potent antihistamines, and two of these compounds displayed moderate anticholinergic activity.

Keyphrases □ Antihistaminic activity—1-[(4-substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines, synthesis, IR, NMR, and mass spectral analyses □ Anticholinergic activity—1-[(4-substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines, synthesis, IR, NMR, and mass spectral analyses □ 1-[(4-Substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines—synthesis, antihistaminic and anticholinergic activity

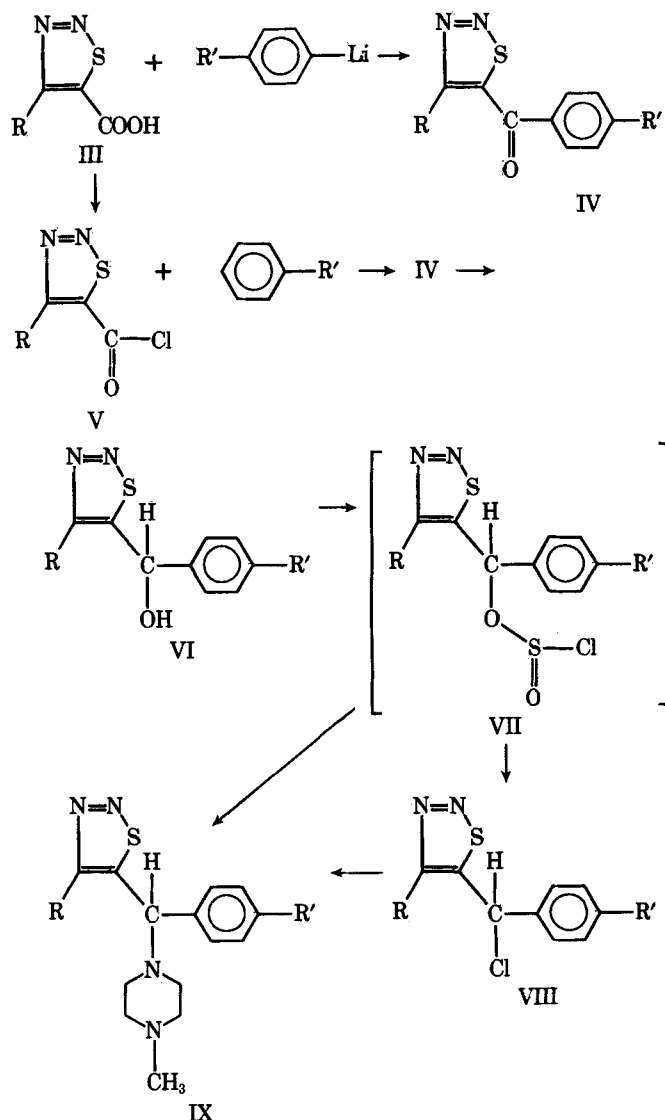
It was reported that piperazine derivatives such as chlorcyclizine (I) and meclizine (II) have significant antihistaminic activity (1). 1,2,3-Thiadiazole derivatives of benzimidazole, benzoxazole, and benzothiazole were reported to be anthelmintics (2). Some phosphorus compounds having the 1,2,3-thiadiazole ring system showed insecticide activity (3), and 4-amino-1,2,3-thiadiazolesulfonamides exhibited antibacterial properties (4). The synthesis and antibacterial activity of 4-substituted-(1,2,3-thiadiazol-5-yl)carbamic acid esters were reported recently (5).

In a continuing effort to find a potent antihistamine with low toxicity (6), a series of 1-[(4-substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines was prepared, and efficacy was determined.

RESULTS AND DISCUSSION

Chemistry—Ethyl 4-substituted-1,2,3-thiadiazole-5-carboxylates were prepared by an oxidative cyclization of methyl or methylene ketone semicarbazones with thionyl chloride (7).

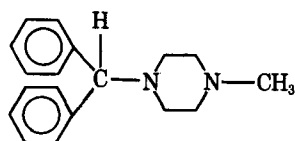
Aryl 4-phenyl-1,2,3-thiadiazol-5-yl ketones (IV, R = C₆H₅) were prepared from the reaction of aryllithium (8) with 4-phenyl-1,2,3-thiadiazole-5-carboxylic acid (III, R = C₆H₅), and aryl 4-methyl-1,2,3-thiadiazol-5-yl ketones (IV, R = CH₃) were prepared from the Friedel-Crafts reaction of 4-methyl-1,2,3-thiadiazole-5-carboxyl chloride (V, R = CH₃)



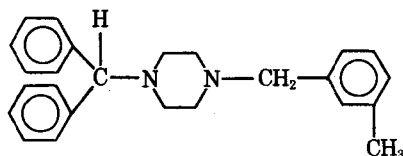
- | | |
|---|---|
| a: R = CH ₃ , R' = H | e: R = CH ₃ , R' = Cl |
| b: R = CH ₃ , R' = CH ₃ | f: R = C ₆ H ₅ , R' = H |
| c: R = CH ₃ , R' = CH ₃ O | g: R = C ₆ H ₅ , R' = CH ₃ |
| d: R = CH ₃ , R' = Br | h: R = C ₆ H ₅ , R' = Cl |

Scheme I

with aromatic hydrocarbones. Reduction of IV with sodium borohydride gave (4-substituted-1,2,3-thiadiazol-5-yl)arylcarbinol (VI). Reaction of VI with thionyl chloride in benzene at room temperature gave VII as an intermediate, which was transformed to VIII at 80°. The latter compound afforded IX through reaction with *N*-methylpiperazine. Transformation of VII (R = CH₃, R' = OCH₃) to VIII could not be achieved under different experimental conditions. In all cases, a mixture of products was obtained. However, VII (R = CH₃, R' = OCH₃) could be converted directly to IX (R = CH₃, R' = OCH₃) through the reaction with *N*-methylpiperazine (Scheme I).



I



II

Table I—Aryl 4-Substituted-1,2,3-thiadiazol-5-yl Ketones^a

Compound	R	R'	Yield, %	Melting Point ^b	Formula	Analysis, %	
						Calc.	Found
IVa	CH ₃	H	70	180–182° ^c	C ₁₀ H ₈ N ₂ O ₂ S	C 58.82 H 3.92 N 13.73	58.66 3.74 13.64
IVb	CH ₃	CH ₃	87	170–172° ^c	C ₁₁ H ₁₀ N ₂ O ₂ S	C 60.55 H 4.59 N 12.84	60.72 4.75 12.99
IVc	CH ₃	OCH ₃	66	53–55°	C ₁₁ H ₁₀ N ₂ O ₂ S	C 56.41 H 4.27 N 11.97	56.60 4.39 11.81
IVd	CH ₃	Br	71	82–84°	C ₁₀ H ₇ BrN ₂ O ₂ S	C 42.40 H 2.47 N 9.89	42.35 2.65 9.72
IVe	CH ₃	Cl	64	55–57°	C ₁₀ H ₇ ClN ₂ O ₂ S	C 50.31 H 2.94 N 11.74	50.50 2.75 11.86
IVf	C ₆ H ₅	H	66	130–132° ^c	C ₁₅ H ₁₀ N ₂ O ₂ S	C 67.67 H 3.76 N 10.53	67.48 3.89 10.65
IVg	C ₆ H ₅	CH ₃	68	88–90°	C ₁₆ H ₁₂ N ₂ O ₂ S	C 68.57 H 4.29 N 10.0	68.71 4.12 10.15
IVh	C ₆ H ₅	Cl	50	137–139°	C ₁₅ H ₉ ClN ₂ O ₂ S	C 59.90 H 2.99 N 9.32	59.98 3.15 9.51

^a IR, NMR, and mass spectra of all compounds were as expected. ^b Unless otherwise indicated, the recrystallization solvent was ether. ^c The boiling point was obtained at 4 mm Hg.

Table II—(4-Substituted-1,2,3-thiadiazol-5-yl)arylcarbinols^a

Compound	R	R'	Yield, %	Melting Point ^b	Formula	Analysis, %	
						Calc.	Found
VIa	CH ₃	H	90	94–96°	C ₁₀ H ₁₀ N ₂ O ₂ S	C 58.25 H 4.85 N 13.59	58.44 4.98 13.74
VIb	CH ₃	CH ₃	84	113–115°	C ₁₁ H ₁₂ N ₂ O ₂ S	C 60.00 H 5.45 N 12.73	59.85 5.63 12.91
VIc	CH ₃	OCH ₃	96	111–113°	C ₁₁ H ₁₂ N ₂ O ₂ S	C 55.93 H 5.08 N 11.86	55.98 5.24 11.98
VI d	CH ₃	Br	72	102–104°	C ₁₀ H ₉ BrN ₂ O ₂ S	C 42.11 H 3.16 N 9.82	42.29 3.28 9.67
VIe	CH ₃	Cl	86	108–110°	C ₁₀ H ₉ ClN ₂ O ₂ S	C 49.90 H 3.74 N 11.64	49.75 3.61 11.46
VI f	C ₆ H ₅	H	91	58–60° ^c	C ₁₅ H ₁₂ N ₂ O ₂ S	C 67.16 H 4.48 N 10.45	67.02 4.29 10.31
VI g	C ₆ H ₅	CH ₃	86	63–64° ^c	C ₁₆ H ₁₄ N ₂ O ₂ S	C 68.09 H 4.96 N 9.93	68.22 4.78 9.75
VI h	C ₆ H ₅	Cl	94	86–88° ^c	C ₁₅ H ₁₁ ClN ₂ O ₂ S	C 59.50 H 3.64 N 9.26	59.28 3.82 9.42

^a IR, NMR, and mass spectra of all compounds were as expected. ^b Unless otherwise indicated, the recrystallization solvent was ether. ^c This compound was crystallized from petroleum ether.

The physical data for the intermediates IV, VI, and VIII and the final compound IX are summarized in Tables I–IV.

Pharmacological Assay—The compounds listed in Table IV were screened for antihistaminic and anticholinergic activities. Isolated guinea pig ileum was prepared according to the method of Magnus (9) and suspended in a 10-ml bath filled with Tyrode solution, which was aerated at 36°. The isotonic concentrations were recorded through an isotonic transducer on a polygraph¹. The pA₂ values were determined by a literature method (10).

Histamine and acetylcholine were used as agonists, and promethazine was tested for comparison. The results are presented in Table V.

Compounds IXe–IXh were the most potent antihistamines. However, only IXe and IXh displayed moderate anticholinergic activity. The LD₅₀ value of IXh in mice, estimated by the moving average method (11), was 155.5 (141.2–171.4) mg/kg.

¹ Narco Biosystems.

EXPERIMENTAL²

4-Phenyl-1,2,3-thiadiazole-5-carboxylic Acid (III, R = C₆H₅)—A solution of ethyl 4-phenyl-1,2,3-thiadiazole-5-carboxylate (23.4 g, 0.1 mole) and sodium hydroxide (4.4 g, 1.1 moles) in ethanol-water (150 ml) was refluxed for 2 hr. The solvent was evaporated, and the residue was crystallized from ethanol to give 15.5 g (75% yield) of III, mp 152–153°.

Anal.—Calc. for C₉H₈N₂O₂S: C, 52.43; H, 2.91; N, 13.59. Found: C, 52.61; H, 2.74; N, 13.76.

Phenyl 4-Phenyl-1,2,3-thiadiazol-5-yl Ketone (IVf)—To a solution of phenyllithium (0.1 mole) in 150 ml of dry ether (12) was added III (R = C₆H₅, 4.12 g, 0.02 mole) at 0°. The mixture was stirred for 36 hr at room

² Melting points were taken on a Kofler hot-stage microscope and are uncorrected. IR spectra were recorded using a Perkin-Elmer 267 spectrometer. Mass spectra were recorded on a Varian Mat III instrument. NMR spectra were determined with a Varian T-60A instrument.

Table III—5-(α -Chlorobenzyl)-4-methyl-1,2,3-thiadiazoles

Compound	R	R'	Yield, %	Melting Point	Formula	Analysis, %	
						Calc.	Found
VIIIa	CH ₃	H	76	138–140° ^a	C ₁₀ H ₉ ClN ₂ S	C 53.45 H 4.01 N 12.47	53.28 4.18 12.65
VIIIb	CH ₃	CH ₃	60	144–146° ^a	C ₁₁ H ₁₁ ClN ₂ S	C 55.35 H 4.61 N 11.74	55.52 4.80 11.92
VIIIc	CH ₃	Br	98	Oil	C ₁₀ H ₈ BrClN ₂ S	C 39.54 H 2.64 N 9.23	39.72 2.46 9.05
VIIIe	CH ₃	Cl	98	140–142° ^a	C ₁₀ H ₈ Cl ₂ N ₂ S	C 46.33 H 3.09 N 10.81	46.48 3.25 10.96
VIIIf	C ₆ H ₅	H	86	104–106° ^b	C ₁₅ H ₁₁ ClN ₂ S	C 62.83 H 3.84 N 9.77	62.95 3.68 9.59
VIIIg	C ₆ H ₅	CH ₃	94	Oil	C ₁₆ H ₁₃ ClN ₂ S	C 63.89 H 4.33 N 9.32	63.71 4.18 9.48
VIIIh	C ₆ H ₅	Cl	95	56–57° ^c	C ₁₅ H ₁₀ Cl ₂ N ₂ S	C 56.07 H 3.12 N 8.72	56.19 3.24 8.91

^a The boiling point was obtained at 4 mm Hg. ^b This compound was crystallized from petroleum ether. ^c This compound was crystallized from ether.

Table IV—[(4-Substituted-1,2,3-thiadiazol-5-yl)arylmethyl]-4-methylpiperazines

Compound	R	R'	Yield, %	Melting Point ^a	Formula	Analysis, %	
						Calc.	Found
IXa	CH ₃	H	62	217–218°	C ₁₅ H ₂₀ N ₄ S	C 62.50 H 6.94 N 19.44	62.68 6.99 19.62
IXb	CH ₃	CH ₃	65	152–155°	C ₁₆ H ₂₂ N ₄ S	C 63.58 H 7.28 N 18.54	63.74 7.45 18.72
IXc	CH ₃	OCH ₃	65	145–148°	C ₁₆ H ₂₂ N ₄ OS	C 60.38 H 6.92 N 17.61	60.19 6.98 17.43
IXd	CH ₃	Br	62	156–160°	C ₁₅ H ₁₉ BrN ₄ S	C 49.05 H 5.18 N 15.26	49.22 5.34 15.08
IXe	CH ₃	Cl	68	160–163°	C ₁₅ H ₁₉ ClN ₄ S	C 55.81 H 5.89 N 17.36	55.69 5.95 17.18
IXf	C ₆ H ₅	H	75	226–230°	C ₂₀ H ₂₂ N ₄ S	C 68.57 H 6.29 N 16.0	68.74 6.18 16.18
IXg	C ₆ H ₅	CH ₃	70	251–255°	C ₂₁ H ₂₄ N ₄ S	C 69.23 H 6.59 N 15.38	69.18 6.74 15.24
IXh	C ₆ H ₅	Cl	86	192–196°	C ₂₀ H ₂₁ ClN ₄ S	C 62.42 H 5.46 N 14.56	62.31 5.29 14.38

^a All compounds were crystallized from absolute ethanol as the hydrochloride.

temperature under nitrogen, water was added, and the ether was evaporated. The residue was distilled to give 3.5 g (66% yield) of IVf, bp 130–132° (4 mm); IR (KBr): 1660 (C=O) cm⁻¹; mass spectrum: *m/z* (relative intensity) 266 (M⁺, 45), 208 (45), 207 (68), 195 (41), 194 (64), 167 (43), 99 (61), 56 (100), and 44 (23).

Anal.—Calc. for C₁₅H₁₀N₂OS: C, 67.67; H, 3.76; N, 10.53. Found: C, 67.48; H, 3.89; N, 10.65.

***p*-Chlorophenyl 4-Phenyl-1,2,3-thiadiazol-5-yl Ketone (IVh)**—To a stirring solution of *p*-chlorophenyllithium, prepared from *p*-chlorobromobenzene (9.575 g, 0.05 mole) and *n*-butyllithium (32 ml of 10% solution in hexane, 0.05 mole) according to the literature (8), was added III (R = C₆H₅, 2.06 g, 0.01 mole). The mixture was stirred overnight under nitrogen at room temperature. Ice water was added to the mixture, followed by extraction with ether. The ether was dried, filtered, and evaporated. The residue was purified by TLC (silica gel, chloroform–petroleum ether, 50:50) and crystallized from ether to give 10.5 g (50% yield) of IVh, mp 137–139°; IR (KBr): 1650 (C=O) cm⁻¹.

Anal.—Calc. for C₁₅H₉ClN₂OS: C, 59.90; H, 2.99; N, 9.32. Found: C, 59.98; H, 3.15; N, 9.51.

Compound IVg was prepared similarly.

4-Methyl-1,2,3-thiadiazol-5-carboxylic Chloride (V, R = CH₃)—A mixture of 4-methyl-1,2,3-thiadiazole-5-carboxylic acid (14.4 g, 0.1 mole) (13) and thionyl chloride (40 ml) was refluxed for 4 hr. The solvent was

evaporated, and the residue was distilled to give 15.5 g (95% yield) of V (R = CH₃), bp 102–104° (20 mm).

Anal.—Calc. for C₄H₃ClN₂OS: C, 29.54; H, 1.85; N, 17.23. Found: C, 29.72; H, 1.98; N, 17.41.

Phenyl 4-Methyl-1,2,3-thiadiazol-5-yl Ketone (IVa)—A stirring mixture of V (R = CH₃, 3.25 g, 0.02 mole) and aluminum chloride (5.34 g, 0.04 mole) in 30 ml of dry benzene was refluxed for 8 hr. After cooling, the complex was decomposed with ice water and dilute hydrochloric acid. The organic layer was separated, and the mother liquor was extracted once more with benzene. The combined organic solvent was washed with a saturated aqueous sodium bicarbonate solution. The organic layer was dried, filtered, and evaporated, and the residue was distilled to give 2.9 g (71% yield) of IVa, bp 180–182° (4 mm); IR (KBr): 1660 (C=O) cm⁻¹; NMR (CDCl₃): 8.0–7.53 (m, 5H, aromatic) and 2.8 (s, 3H, CH₃) ppm; mass spectrum: *m/z* (relative intensity) 204 (M⁺, 2), 186 (38), 105 (28), and 77 (100).

Anal.—Calc. for C₁₀H₈N₂OS: C, 58.82; H, 3.92; N, 13.73. Found: C, 58.66; H, 3.74; N, 13.64.

Compounds IVb–IVe were prepared similarly at the boiling point of the respective solvent, except IVc which was prepared at 100° (Table I).

(4-Methyl-1,2,3-thiadiazol-5-yl)phenylcarbinol (VIa)—To a stirring solution of IVa (2.04 g, 0.01 mole) in 50 ml of methanol was added

Table V—Antihistaminic and Anticholinergic Activities of 1,2,3-Thiadiazole Derivatives

Compound	pA ₂ Values	
	Antihistaminic Activity	Anticholinergic Activity
IXa	5.37	NT ^a
IXb	4.59	NT ^a
IXc	5.02	— ^b
IXd	5.22	— ^b
IXe	6.69	4.94
IXf	6.92	— ^b
IXg	6.67	— ^b
IXh	7.72 ± 0.26 (5)	4.6
Promethazine	10.47	6.9

^a Not tested. ^b Inactive up to the concentration of $2 \times 10^{-5} M$.

sodium borohydride (0.38 g, 0.01 mole). The mixture was stirred 30 min. Water (100 ml) was added to the mixture, which then was extracted with chloroform (3 × 100 ml). The chloroform was dried, filtered, and evaporated, and the residue was crystallized from ether to give 1.85 g (90% yield) of VIa, mp 94–96°; IR (KBr): 3250 (OH) cm^{-1} ; NMR (CDCl₃): 7.36 (s, 5H, C₆H₅), 6.11 (s, 1H, HCO), 3.56 (broad s, 1H, OH), and 2.41 (s, 3H, CH₃) ppm; mass spectrum: *m/z* (relative intensity) 206 (M⁺, 1), 177 (22), 107 (100), 79 (99), 77 (99), 51 (66), and 45 (65).

Anal.—Calc. for C₁₀H₁₀N₂OS: C, 58.25; H, 4.85; N, 13.59. Found: C, 58.44; H, 4.98; N, 13.74.

Compounds VIb–VIIh were prepared similarly (Table II).

5-(α -Chlorobenzyl)-4-methyl-1,2,3-thiadiazole (VIIIa)—A solution of VIa (2.06 g, 0.01 mole) and thionyl chloride (4 ml) in 120 ml of dry benzene was stirred overnight. The mixture was filtered, and the solvent was evaporated. The residue was distilled to give 1.70 g (76% yield) of VIIIa, bp 138–140° (4 mm); NMR (CDCl₃): 7.40 (s, 5H, C₆H₅), 6.25 (s, 1H, HCCl), and 2.55 (s, 3H, CH₃) ppm; mass spectrum: *m/z* (relative intensity) 224 (M⁺, 2), 195 (56), 161 (56), 125 (68), and 59 (100).

Anal.—Calc. for C₁₀H₉ClN₂S: C, 53.45; H, 4.01; N, 12.47. Found: C, 53.28; H, 4.18; N, 12.65.

Compounds VIIIb–VIIIh were prepared similarly (Table III).

1-[(4-Methyl-1,2,3-thiadiazol-5-yl)phenylmethyl]-4-methylpiperazine (IXa)—A solution of VIIIa (2.245 g, 0.01 mole) and *N*-methylpiperazine (2.0 g, 0.02 mole) in 30 ml of pyridine was refluxed under nitrogen overnight. The solvent was evaporated. Water (15 ml) was added to the residue and then extracted with ether. The ether was evaporated, and the residue was purified by TLC (chloroform–methanol, 95:5) to give 1.73 g (60% yield) of IXa; NMR (CDCl₃): 7.33 (s, 5H, C₆H₅), 4.7 (s, 1H, HCN), 2.56 (s, 3H, CH₃), 2.46 (s, 8H, CH₂N), and 2.3 (s, 3H, CH₃) ppm.

Anal.—Calc. for C₁₅H₂₀N₄S: C, 62.50; H, 6.94; N, 19.44. Found: C, 62.68; H, 6.99; N, 19.62.

This compound was crystallized as the hydrochloride (mp 217–218°) from absolute ethanol.

Compounds IXb and IXd–IXh were prepared similarly (Table IV).

1-[(4-Methyl-1,2,3-thiadiazol-5-yl)-*p*-methoxyphenylmethyl]-4-methylpiperazine (IXc)—A solution of VIc (2.36 g, 0.01 mole) and thionyl chloride (4 ml) in 120 ml of benzene was stirred overnight. The solvent was evaporated under reduced pressure at 0° to give VIc as an oil; IR (KBr): 1250 and 1180 (SO₂) cm^{-1} ; NMR (CDCl₃): 7.14 (q, 4H, aromatic), 6.26 (s, 1H, HCOSOC), 3.8 (s, 3H, CH₃O), and 2.55 (s, 3H, CH₃) ppm. This compound was not purified further. To the residue were added pyridine (25 ml) and *N*-methylpiperazine (2 g, 0.02 mole). The mixture was refluxed under nitrogen and treated as was IXa to give 2.1 g (66% yield) of IXc; NMR (CDCl₃): 7.0 (q, 4H, aromatic), 4.65 (s, 1H, HCN), 3.73 (s, 3H, CH₃O), 2.53 (s, 3H, CH₃), 2.43 (s, 8H, CH₂N), and 2.23 (s, 3H, CH₂N) ppm.

Anal.—Calc. for C₁₆H₂₂N₄O₂S: C, 60.38; H, 6.92; N, 17.61. Found: C, 60.19; H, 6.98; N, 17.43.

This compound was crystallized from absolute ethanol as the hydrochloride, mp 145–148°.

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ACKNOWLEDGMENTS

Supported by Grant 400-7-35/2 from the Ministry of Culture and Higher Education.